

1 TITLE:

2 Preparation of keratin hydrolysate from chicken feathers and its application in cosmetics.

3

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13

14 KEYWORDS:

15 alkaline-enzyme hydrolysis; chicken feathers; keratin; keratin hydrolysate; cosmetic
16 formulation; ointment base; humectant; hydration; transepidermal water loss

17

18 SHORT ABSTRACT:

19 The goal of the protocol is to prepare keratin hydrolysate from chicken feathers by alkaline-
20 enzymatic hydrolysis and to test whether adding of keratin hydrolysate into a cosmetics
21 ointment base improves skin barrier function (heighten hydration and diminishing
22 transepidermal water loss). Tests are conducted on men and woman volunteers.

23

24 LONG ABSTRACT:

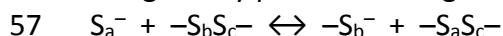
25 Keratin hydrolysates (KH) have established as standard components in hair cosmetics.
26 Understanding the moisturising effects of KH is advantageous for skin-care cosmetics. The goals
27 of the protocol are: (1) to process chicken feathers into KH by alkaline-enzymatic hydrolysis and
28 purify it by dialysis, and (2) to test if adding KH into an ointment base (OB) increases hydration
29 of the skin and improves skin barrier function - diminishes transepidermal water loss (TEWL).
30 During alkaline-enzymatic hydrolysis feathers are firstly incubated at higher temperature in
31 alkaline environment and then, under mild conditions, hydrolyzed with proteolytic enzyme.
32 Solution of KH is dialyzed, vacuum dried and milled to a fine powder. Cosmetic formulations
33 comprising from oil in water emulsion (O/W) containing 2, 4 and 6 weight % of KH (based on
34 weight of the OB) are prepared. Testing the moisturizing properties of KH is carried out on 10
35 men and 10 women at time intervals of 1, 2, 3, 4, 24 and 48 h. Tested formulations are spread
36 at degreased volar forearm sites. Skin hydration of stratum corneum (SC) is assessed by
37 measuring capacitance of the skin, which is one of the most world-wide used and simple
38 method. TEWL is based on measuring the quantity of water being transported per a defined
39 area and period of time from the skin. Both methods are fully non-invasive. KH makes for an
40 excellent occlusive; depending on the addition of KH into OB it brings about a 30 % reduction in
41 TEWL after application. KH also functions as a humectant, as it binds water from the lower
42 layers of the epidermis to the SC; at optimum KH addition into the OB up to 19 % rise in
43 hydration in men and 22 % rise in women occurs.

44

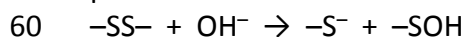
45 **INTRODUCTION:**

46 Slaughterhouses, food industry and tanning industry annually produce immense amounts of
47 solid keratin by-products – wool, feathers, bristles, hooves, claws, horns and the like. According
48 to latest statistical data total live weight of chickens, turkeys, ducks and other slaughtered
49 poultry in the USA is **62.5 billion** pounds per year¹; in the EU it is approx. **28.7 billion** pounds per
50 year. Considering that feathers make up to 8.5 % of total poultry weight, only the USA itself
51 annually produces approx. **5.3 billion** pounds of waste feathers².

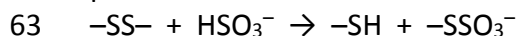
52
53 Keratin is a protein exhibiting high chemical resistance because it is strongly cross-linked with
54 disulfide bridges that render its processing difficult. Obtaining soluble products requires
55 cleaving cross-links and possibly carrying out hydrolysis of peptide bonds³. Cleavage of disulfide
56 bridges may proceed through a reaction of thiol anion according to following pattern^{4, 5}:



58 With a very high pH level, hydrolysis of disulfide bridges also appears, in accordance with
59 pattern⁶:



61 Under mild conditions (pH approx. 8), even **sulfitolysis** takes place according to following
62 pattern:



64
65 The most economical way of degrading keratin is microbial breakdown which is characterized,
66 by mild reaction conditions during processing and high breakdown efficiency (approx. 90 %)^{7, 8}.
67 Keratinases are produced by some bacteria isolated from soil and keratin waste⁹. Microbial
68 keratinases **hydrolyze** rigid and strongly cross-linked keratin structures¹⁰ and keratin
69 hydrolysate prepared is rich in soluble proteins and no loss in essential amino acids was
70 detected with it¹¹.

71
72 In order to incorporate a protein in cosmetic preparations (e.g. emulsions, lotions and gels),
73 requirements ensure such proteins are soluble in water, the given systems are transparent and
74 that re-aggregation of peptides is avoided due to hydrophobic interactions. Therefore, a
75 common practice is to apply hydrolysates of proteins, such as hydrolyzed collagen, elastin and
76 keratin. When adding hydrolysates into cosmetic emulsions, steps are taken to ensure that the
77 hydrolysate is firstly dissolved in water. In some cases it is desirable that the protein - or the
78 hydrolysate - is soluble in alcohol or other organic solvents¹².

79
80 KH normally features in shampoos, conditioners, lotions, and nutritive serums for hair, as well
81 as mascaras, nail polish and eye make-up agents. The KH effects declared usually include
82 forming a protective film, smoothing the hair or nail structure, heightened plasticity and
83 appearance of the treated formation, regulating the consistency of products and encouraging
84 the formation of foam^{13, 14}. It has also been shown that KH reduces surface tension, hence
85 supplementation in cosmetics can facilitate reduction in the amount of emulsifier added to
86 stabilize creams. KH limit the effects of irritation triggered by cleaning agents (surfactants) to
87 the skin, eyes and hair, thus reducing any potential side effects of cleaning agents on tissue (e.g.
88 dehydration of the skin, hardness and decreased barrier function of the skin). The high

89 buffering capability of hydrolysates is also exploited to stabilize the pH of cosmetics; peptides of
90 shorter length have a greater buffering effect^{15, 16}. Although KH have become established as
91 standard components in hair and nail cosmetics as well as being utilized in products for skin
92 care, studies on the moisturising effects of KH do not appear among contemporary literature.

93
94 At our workplaces, alkaline-enzymatic technology has been developed for processing keratin
95 by-products into KH, and active testing is in process on the effects of a number of cosmetic
96 additives¹⁷⁻²². The advantage of 2-stage alkaline-enzymatic hydrolysis using microbial proteases
97 for chicken feathers achieves high efficiency under mild reaction conditions and the quality of
98 KH is very high in contrast to hydrolysis employed in strong acids or alkalis. In the first stage,
99 feathers are incubated at higher temperature in alkaline environment which partially disrupt
100 the keratin structure and swell the feathers; after adjusting pH the feathers are hydrolyzed with
101 proteolytic enzyme under mild conditions in the second stage. Dialyzed KH possesses a high
102 content of proteins.

103
104 The purposes of this method we describe here are processing poultry feathers into a KH
105 through alkaline-enzymatic hydrolysis and testing the effect of moisturising properties of KH
106 applied to O/W cosmetic emulsion. The moisturising properties are investigated by
107 instrumental non-invasive methods *in vivo*. The most frequent methods for measuring skin
108 hydration and barrier function of SC include measuring electrical properties of the skin
109 (conductance or capacitance). Different methods for investigating SC hydration include near
110 infrared multispectral imaging method (NIM), nuclear magnetic resonance spectroscopy,
111 optical coherence tomography or transient thermal transfer²³. Barrier function of SC correlates
112 to TEWL of SC and it is measured by ventilated chamber method, unventilated chamber
113 method and open chamber method²⁴.

114
115 Properties of model formulations are determined using Multi Probe adapter MPA 5 with three
116 types of probes. The first one, corneometer CM 825, measures skin hydration by assessing
117 changes in the electrical capacity of the skin's surface; the measuring capacitor shows changes
118 in capacitance of the skin surface in corneometric units. The corneometer gives only a relative
119 assessment of skin hydration²⁵. For TEWL, the second probe, tewameter TM 300 is used for
120 measuring the density gradient of water evaporation (open chamber instrument based on Fick's
121 diffusion law) from the skin indirectly by the two pairs of sensors (temperature and relative
122 humidity) indicating the quantity of water being transported per a defined area and period of
123 time (g/m²/h). This method can detect even the slightest disruption of skin barrier function²⁶.
124 Skin pH is one of the indicators of barrier and anti-microbial function of SC²⁷. The acidity of the
125 skin mantle was measured by a (third) skin PH 905 probe connected to MPA 5 station. This
126 specially designed probe consists of a flat-topped glass electrode for full skin contact,
127 connected to a voltmeter. The system measures potential changes due to the activity of
128 hydrogen cations surrounding the very thin layer of semi-solid forms measured at the top of the
129 probe. The changes in voltage are displayed as pH²⁸.

130
131 We present experiments divided into three sections. (1) Preparation of KH from chicken
132 feathers by two-stage alkaline-enzymatic hydrolysis and its purification by dialysis (removing

133 salts and low-molecular fractions). (2) Preparation of cosmetic formulations containing 2, 4 and
134 6 % KH. (3) Testing the properties of KH by measuring skin hydration, TEWL and skin pH. Testing
135 was carried out on 10 women with the mean age of 27.2 years and on 10 men with the mean
136 age of 26.2 years. The method of selecting the volunteers and the testing itself were conducted
137 in accordance with international ethical principles of bio-medical research utilising human
138 subjects²⁹; all persons gave their informed consent prior to inclusion in the study. Before testing
139 commenced, the volunteers were asked to complete a questionnaire on their health status. The
140 volunteers committed to avoid applying any cosmetic product to the test sites and surrounding
141 regions during the 24 h prior to and during the test period; furthermore, they were only
142 permitted brief evening washes with running water.

143

144 **PROTOCOL:**

145

146 **1. Process chicken feathers into keratin hydrolysate**

147 Collect chicken feathers from a poultry farm.

148

149 1.1. Wash out any insoluble impurities and blood remnants from the chicken feathers with a
150 sufficient excess of fresh running (cold) water. Place the feathers on a flat plate and dry
151 overnight at 50 °C.

152 **Note:** The protocol can be paused here.

153

154 1.2. Grind the dried feathers in a cutting mill (suitable for soft to medium-hard sample
155 materials, and fibrous materials) into final fineness 1.0 mm. Alternatively, final fineness of
156 grinded feathers could be higher, but not more than 3.0 mm.

157 **Note:** The protocol can be paused here.

158

159 1.3. Degrease feathers

160 The most effective and economic method of degreasing poultry feather is using a commercial
161 lipolytic enzyme.

162

163 1.3.1. In a stainless steel 27-L boiler container with temperature control mix feathers with
164 water preheated up to 40±2 °C in a weight ratio 1:75, add a lipolytic enzyme (activity 100
165 KLU/g) in a dose of 1.5-2.0 % (related to weighed-in dry feathers) and gently stir the content
166 with overhead stirrer for 5 min.

167

168 1.3.2. Adjust the mixture pH to the level corresponding to the maximum activity of the lipolytic
169 enzyme by adding 1-% NaOH or 1-% H₃PO₄ solution. In this case adjust pH to 9.0±0.2 by adding
170 1-% NaOH solution. Stir the mix for 5 min, check and re-adjust the pH level.

171

172 1.3.3. Gently stir the mixture with overhead stirrer for 24 h at 40±0.5 °C. Alternatively, incubate
173 the mixture at 40±0.5 °C and during the first 6 h of incubation shortly stir the content in 1-hour
174 intervals.

175

176 1.3.4. Filter the mixture through a fine sieve (100 µm size) and wash degreased feathers with a

177 stream of fresh running (cold) water and dry the feathers on a flat plate at 50 °C in chamber
178 drier overnight.

179 **Note:** The protocol can be paused here.

180

181 1.4. Perform the first stage of chicken feathers hydrolysis. Mix feathers with 0.3-% KOH water
182 solution in a weight ratio 1:50 and gently stir with overhead stirrer at 60±0.5 °C for 24 h. Level
183 of pH in the mixture decreases from approx. 12.5 at the start of incubation to approx. 11.0 at
184 the end of incubation. On finishing first stage of hydrolysis, adjust mixture pH to the level
185 corresponding to the maximum activity of the proteolytic enzyme with 10-% H₃PO₄ (in this case
186 to a level of 9.0±0.2).

187

188 1.5. Perform the second stage of chicken feathers hydrolysis. Add to the mixture **proteolytic**
189 **enzyme** in a dose of 5.0 % (related to dry matter of quantity of feathers weighed-in at the
190 beginning of the first stage of hydrolysis). Gently stir with overhead stirrer at 60±0.5 °C for 8 h
191 and then heat the mixture (**in the same stainless steel 27-L boiler container**) to a boiling point
192 (**100 °C**) and boil for 10 min to inactivate enzyme.

193

194 1.6. Separate solution of KH (**prepared in step 1.5**) from undissolved remnant by filtering it
195 through low-density filter paper on a Buechner funnel with slight under-pressure; alternatively
196 use a centrifuge.

197 **Note:** The protocol can be paused here for several days if a solution of KH is stored at 5±1 °C.

198

199 1.7. **In the plastic bucket (26 cm in diameter and 26 cm high) dialyze keratin hydrolysate using**
200 **12K MWCO membrane to remove small peptides and salts.** Pour **400 mL of** solution of KH into
201 dialysis tubing and dialyse it against **4 L of** distilled water for 80 h at room temperature; after
202 18, 36 and 60 h change distilled water.

203

204 1.8. Cast a dialysed solution of KH on an anti-adhesive plate (e.g. silicone one) **on ratio of**
205 **500 mL to 1,000 cm² plate area**, vacuum dry it on a thin film at 40±0.5 °C for overnight, mill to
206 form a fine powder and keep it in a closed vessel in a desiccator.

207 **Note:** The protocol can be paused here for several months if KH powder is stored in a dry place.

208

209 **2. Prepare cosmetic formulations with keratin hydrolysate**

210 **Note:** The ointment base used for testing was an **commercial hydrophilic oil in water (O/W)**
211 **cream base** and comprised of aqua, paraffin, paraffin liquid, cetearyl **alcohol**, Laureth 4, sodium
212 hydroxide, carbomer, methylparaben, propylparaben.

213

214 2.1. Prepare formulations containing 2, 4 and 6 % KH (in accordance with the base weight of the
215 ointment). Weigh the amount of KH powder into a polyethylene vessel (7 cm in diameter and
216 10 cm high) and add the ointment base at an amount that ensured the total weight of the
217 formulation equalled 50 g; **see schedule in Table 1.**

218

219 **[Place Table 1 here]**

220

221 2.2. Homogenize the mixture with a 3-bladed laboratory stirrer for 10 min at 134.16 g using a
222 mechanical overhead stirrer. Maintain the prepared formulations at 5 ± 1 °C and condition them
223 at room temperature for 2 h prior to use.

224 **Note:** The protocol can be paused here for up to 5 months if formulations are stored at 5 ± 1 °C.
225

226 **3. Test the properties of KH by measuring skin hydration, TEWL and pH**

227 **Note:** Perform all measurements in a conditioned room at 23 ± 2 °C and the relative humidity of
228 56 ± 3 %.

229
230 3.1. Place 5 strips of filter paper (size 2x4 cm) into the physiological solution (0.90 % NaCl) and
231 left them for approximately 1 min in the solution.

232
233 3.2. Apply two strips to the inner side of the right forearm, and three to the inner side of the
234 left forearm and fix them with adhesive plasters. This step is aimed to degrease the skin and
235 eliminate individual characteristics of the skin at the site. After 4 h, remove the strips and
236 demark the areas with a permanent pen, see Figure 1.

237
238 [Place Figure 1 here]

239
240 3.3. Apply 0.1 mL of the tested formulations at each spot of degreased forearm sites using
241 syringes and spread it over the entire marked surface. On the left arm, left the first site clear (it
242 is control), the OB apply to the second site and OB + 2 % KH put on the third area. OB + 4 % KH
243 and OB + 6 % KH apply to the right arm.

244
245 3.4. Measure each sample at each site and each interval (1, 2, 3, 4, 24, and 48 h) and take 5
246 readings with corneometer CM 825 probe for skin hydration, 15 readings with tewameter TM
247 300 probe for skin TEWL and 1 reading with pH 905 probe for skin pH. Do not allow volunteers
248 to apply any cosmetic product to the test sites and surrounding areas during the test period;
249 they can only permit brief evening washes with running water.

250 **Note:** The protocol can be paused here.

251
252 3.5. Process the resultant readings by basic numerical characteristics of the descriptive
253 statistics, using table processor. From the 5 hydration readings measured for each sample
254 ignore the lowest and the highest readings and calculate only three readings for arithmetic
255 mean and standard deviation. From the 15 TEWL readings measured for each sample ignore the
256 first 5 and calculate only 10 readings for arithmetic mean and standard deviation.

257 258 **REPRESENTATIVE RESULTS:**

259 KH prepared according to procedure presented in Protocol 1 (see Figure 2) is yellow in color,
260 easily soluble in water with high protein content (inorganic solids represent < 2.0 %); pH of
261 1.0% solution of KH is 5.3 and fulfils requirements for cosmetic-grade hydrolysates. Yield of KH
262 from raw material is approx. 30 %. The molecular weight distribution of KH was determined by
263 SDS-PAGE and is shown in Figure 3.

264

265 [Place Figures 2 and 3 here]

266

267 The hydration and TEWL values as measured were delineated in different volunteers, thus
268 could not be compared to each other. Therefore, the values are expressed as change in per
269 cent in comparison with the OB on the site treated with the formulations, the latter containing
270 2, 4 and 6 % additions of KH at intervals for measurement of 1, 2, 3, 4, 24 and 48 h. The pH
271 values for skin are expressed as the arithmetic mean of the recorded values of skin pH for all
272 volunteers. The results for alteration in hydration and TEWL in per cent, relative to the OB, and
273 for pH levels of the skin are given in Table 2 for 10 men volunteers and in Table 3 for 10 woman
274 volunteers.

275

276 [Place Tables 2 and 3 here]

277

278 **Hydration of SC for men volunteers.**

279 At short intervals of measurement (1-4 h) it is clear that the highest increases in hydration of
280 the skin (12-19 %) were recorded for formulations supplementing the OB with 2 % of KH; the
281 addition of 4 % KH showed a lesser rise (1-7 %) in hydration. Conversely, KH at 6 % reflected
282 negatively on values for SC hydration (a decrease of 3-7 %). After 24 h of measurement, an
283 increase of 11 % in SC hydration was discerned for all the additions of KH to the OB tested. The
284 same trend continued even after 48 h, following with a slight increase in hydration still
285 remained: a 15 % increase for KH at 2 %, a rise of 14 % for the 4 % addition of KH, and the drop
286 of 17 % for KH at 6 %.

287

288 **Hydration of SC for women volunteers.**

289 It is observable that supplementing the OB with 2 % KH causes an approximate 22 % increase in
290 SC hydration, relative to OB alone, as early as at 1 h of measurement; a 4 % addition of KH to
291 the OB has, in fact, no effect on hydration; while, conversely, adding 6 % KH to the OB is
292 reflected in an approximate 12 % decrease in hydration, as compared with pure OB. Similar
293 tendencies are seen after 2, 3 and 4 h of measurement, at which hydration increases by
294 12-15 % for supplementation with 2 % KH; for larger additions of KH, hydration either stays the
295 same or diminishes. After 24 h of measurement, hydration was recorded as higher than for the
296 OB, this for all the tested KH additions; the greatest increase (14 %) in hydration occurred for
297 the addition of 2 % KH, while the lowest increase (8 %) was seen for supplementation at 6 % of
298 KH. Similar results are achieved after 48 h of measurement, wherein greater hydration than for
299 pure OB was recorded for all samples with additions of KH; the biggest rise (18 %) in hydration
300 occurred for the 2 % KH addition, while the lowest increase (10 %) was noted for KH
301 supplementation at 6 %.

302

303 **TEWL for men volunteers.**

304 The results of TEWL make it clear that formulations supplemented with KH diminish TEWL,
305 when applied to the skin, in comparison with pure OB. A heightened amount of KH exerted a
306 positive effect on lower values of TEWL. An hour after applying the formulations, TEWL was
307 recorded as 20 % lower than for pure OB, pertaining to a formulation with 2% KH; while KH at
308 4 % became manifest as a drop in TEWL of 28 %; whereas KH at 6 % resulted in a dramatic 36 %

309 decrease in TEWL. Indeed, diminished TEWL values were also observed at 2, 3 and 4 h of
310 measurement for formulations supplemented with KH. After 24 and 48 h, TEWL was even
311 significantly lowered on sites treated with the KH formulations. After 24 h, TEWL observed on
312 the skin was 23 % lower for the OB with 2 % addition of KH than at the site for pure OB; KH at
313 4 % brought about a drop in TEWL by about 47 %, while KH at 6 % triggered TEWL to fall by
314 53 %. A similar trend is evident even after 48 h – the TEWL for skin treated with OB containing
315 2 % KH was 21 % lower than at the site for pure OB; KH at 4 % showed a TEWL at 36 % lower;
316 and KH at 6 % brought about a drop of 54 %.

317

318 **TEWL for women volunteers.**

319 Similarly to men volunteers, it is evident that all the monitored additions of KH to OB shall be
320 reflected in reduced TEWL. After 1 h of measurement, a significant decrease in TEWL was
321 recorded for OB samples containing all the additions of KH; about a 32 % reduction in TEWL was
322 seen for the 2 % addition of KH, around a 41 % decrease for 4 % KH, and even a 50 % drop in
323 TEWL for 6 % supplementation of KH. Following 2, 3 and 4 h of measurement, the situation
324 remains similar, i.e. there is a decrease in TEWL at such time intervals; the least drop in TEWL
325 occurs for 2 % KH, while the greatest is seen for 6 % KH. After 2 h, TEWL decreases by 16 % for
326 KH at 2 %, 37 % for KH at 4 % and 39 % for KH at 6 %. At 3 h, TEWL diminishes further by 12 %
327 for KH at 2 %, by 24 % for KH at 4 % and 29 % for KH at 6 %. At 4 h, TEWL further decreases by
328 20 % for 2 % KH, 34 % for 4 % KH, and 39 % for 6 % KH. After 24 h, the least reduction in TEWL
329 (16 %) occurred for the 6 % addition of KH, while the greatest (44 %) was seen for 4 %
330 supplementation of KH; KH at 2 % was observed to cause a 35 % decrease in TEWL. At 48 h, the
331 results stay similar and relatively balanced - the least decrease in TEWL (33 %) for the 6 %
332 addition of KH, the greatest (38 %) for 4 % KH and for 2 % KH.

333

334 **FIGURE AND TABLE LEGENDS:**

335 **Figure 1: Method for location of test formulations on the forearm of the left and right upper**
336 **limbs.**

337 **Figure 2: Representative picture of keratin hydrolysate.**

338 **Figure 3: SDS-PAGE of keratin hydrolysate and protein standards; lane 1-ultra low range**
339 **molecular weight Marker (3.5-26.6 kDa), lanes 2, 3 and 4-keratin hydrolyates prepared in 3**
340 **batches, lane 5-wide range molecular weight Marker (6.5-200 kDa)**

341 **Table 1: Weight-in quantities of ointment base and keratin hydrolysate to prepare cosmetic**
342 **formulations.**

343 **Table 2: Results for change in hydration and TEWL and pH of skin of 10 men volunteers at the**
344 **measurement intervals of 1, 2, 3, 4, 24 and 48 h.**

345 **Table 3: Results for change in hydration and TEWL and pH of skin of 10 women volunteers at**
346 **the measurement intervals of 1, 2, 3, 4, 24 and 48 h.**

347

348 **DISCUSSION:**

349 The advantage of alkaline-enzymatic hydrolysis is that it can be modified according to future
350 applications of KH. For example, in hair-care cosmetics applications where a lightly brownish
351 color of a product is not an obstacle a higher temperature in the hydrolysis can be applied
352 leading to higher yield of KH. In addition, longer processing time during both stages of the

353 technological procedure significantly affects overall process efficiency – yield of KH rises up to
354 85 %.

355
356 Findings for hydration measurement make it evident that, during the monitored interval of
357 measurement (1-48 h), the addition of 2 % KH to OB is optimum since it causes a 11-19 %
358 increase in SC hydration for men volunteers, respectively a 12-22 % increase for woman
359 volunteers. The KH that is added possesses a broad distribution of molecular weight. **We**
360 **suggest** that a portion of low-molecular-weight fractions ($M_r < 20$ kDa) penetrates the
361 epidermis following application to the skin. Increased hydration of the skin after putting on a
362 formulation supplemented with KH is explained through KH binding water from lower
363 epidermis layers to the structure of the SC, leading to the formation of H-bridges between KH
364 molecules and water. This mechanism of action is also embraced by some authors³⁰. The
365 moisturising effect of KH is comparable with conventional moisturisers (e.g. glycerol, urea and
366 hyaluronic acid) that were tested in emulsion and gel formulations²².

367
368 TEWL measurements highlight that, during the observed time of measurement (1-48 h) at men
369 volunteers, all the formulations with KH supplementation caused TEWL to decline after
370 application. The 2 % addition of KH to the OB caused a reduction in TEWL of 11-23 %, in
371 comparison with pure OB. When the OB was supplemented with 4 % KH, TEWL dropped by
372 28-47 %; while for KH at 6 % it diminished by as much as 31-54 %. For woman volunteers
373 supplementing the OB with 4 % KH represents the best option, as there is 24-44 % decrease in
374 SC TEWL. The significantly lower TEWL for formulations supplemented with KH can be
375 explained by the process of higher-molecular-weight fractions of KH forming a protective film
376 once applied to the epidermis, thereby preventing the loss of epidermal water. In fact, the
377 highly positive effect of KH on TEWL is comparable or even exceeds, for the sake of comparison,
378 values of TEWL recorded for cosmetic gels or emulsions supplemented with 5-10 % of glycerol
379 and 1-5 % of sericin. Then same is in comparison with conventional mineral oils that diminish
380 TEWL by approximately 25-30 %³¹. Additionally, the barrier properties of KH are better than, for
381 example, those for urea and hyaluronic acid²².

382
383 pH of the skin surface. For reference, pH 3.5 to 4.3 is an acidic skin surface, pH 4.4 to 5.5 is
384 neutral in this respect, and pH 5.6 to 6.5 represents a basic skin surface³². We found that no
385 significant changes were observed in the pH of the skin surface after putting on all of the tested
386 formulations (OB + 2, 4 and 6 % KH); the pH of 4.6 to 5.0 (men volunteers), respectively 4.9 to
387 5.4 (woman volunteers) corresponds to a normal skin surface.

388
389 **Longer-term analysis (more than 2 days) was not accomplished.**

390
391 **Modifications and troubleshooting: Processing of chicken feathers into KH is very easy, runs**
392 **under atmospheric pressure and at mild temperature; the process can be favorably**
393 **transformed from a laboratory scale to a pilot plant scale and an industrial scale. In the 2nd step**
394 **of the protocol where KH is homogenized with the (O/W) emulsion base some modifications**
395 **are possible. In industrial practice O/W and W/O emulsions are prepared by mixing water (W)**
396 **phase (water + cosmetic ingredients soluble in water) and oil (O) phase (oil + cosmetic**

397 ingredients soluble in oil). KH is soluble in water, so it is favorable to blend it directly into water
398 phase of the system.

399

400 Limitations of the technique: localization of each spot for measuring hydration and TEWL and
401 hammered pressure at measuring with corneometer skin probe.

402

403 Critical steps within the protocol lies above all in the 3rd step of the protocol (Testing the
404 properties of KH by measuring skin hydration, TEWL and pH). Health state, individual
405 differences, smokers/non-smokers, gender, age differences, menstruation and mental
406 condition can influence measuring of skin hydration and TEWL. For acquiring representative
407 results the same person should apply tested formulations on forearms and measure hydration
408 and TEWL values. It is vital to perform all measurements in a conditioned room with a stable
409 temperature and relative humidity. In case of measuring values at intervals of 24 and 48 h
410 acclimatization of volunteers in a conditioned room for at least 30 min prior to measurement is
411 necessary.

412

413 **ACKNOWLEDGMENTS:**

414 This article was written with support of the project IGA/FT/2017/007 of Tomas Bata University
415 in Zlin.

416

417 **DISCLOSURES:**

418 The authors have nothing to disclose.

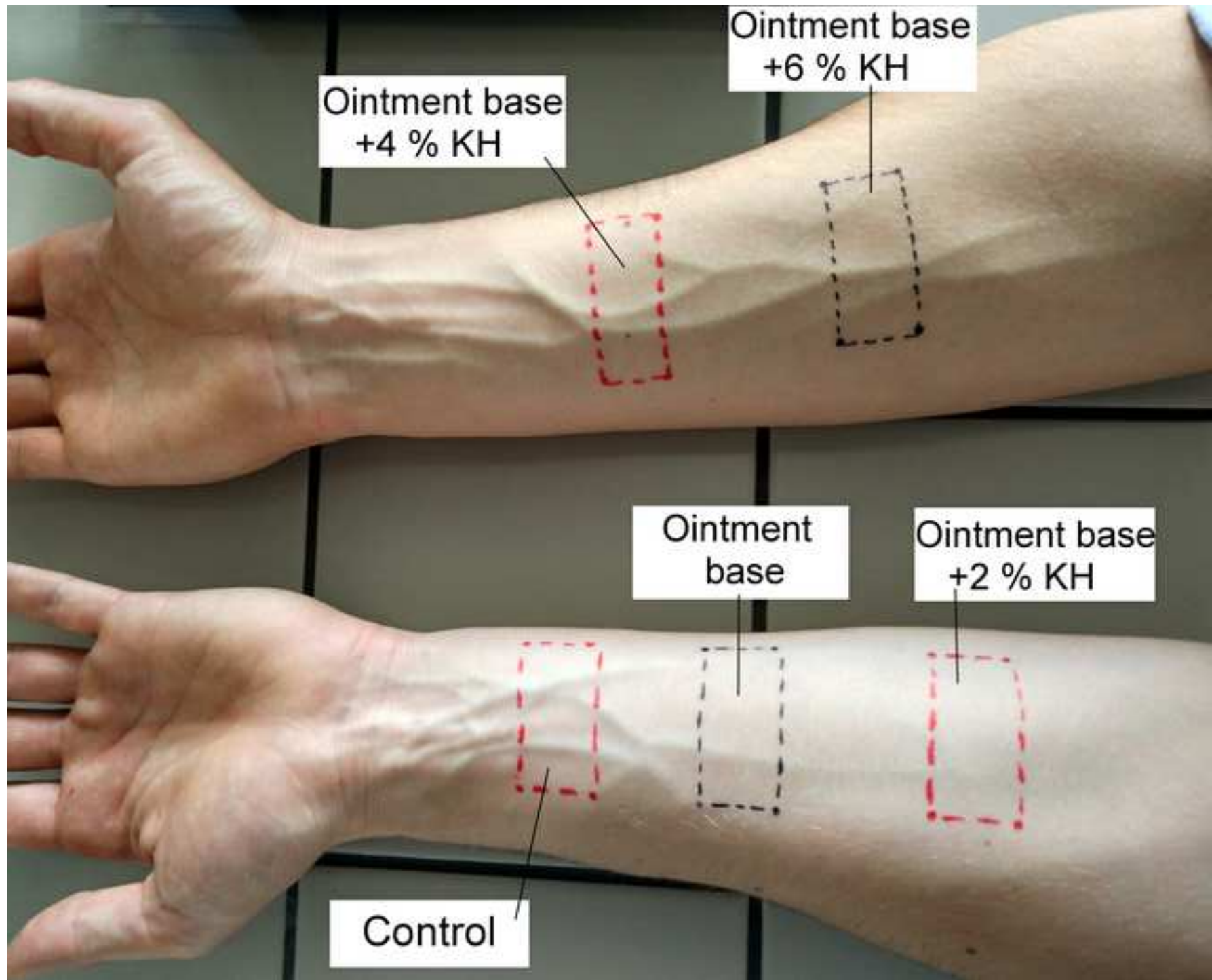
419

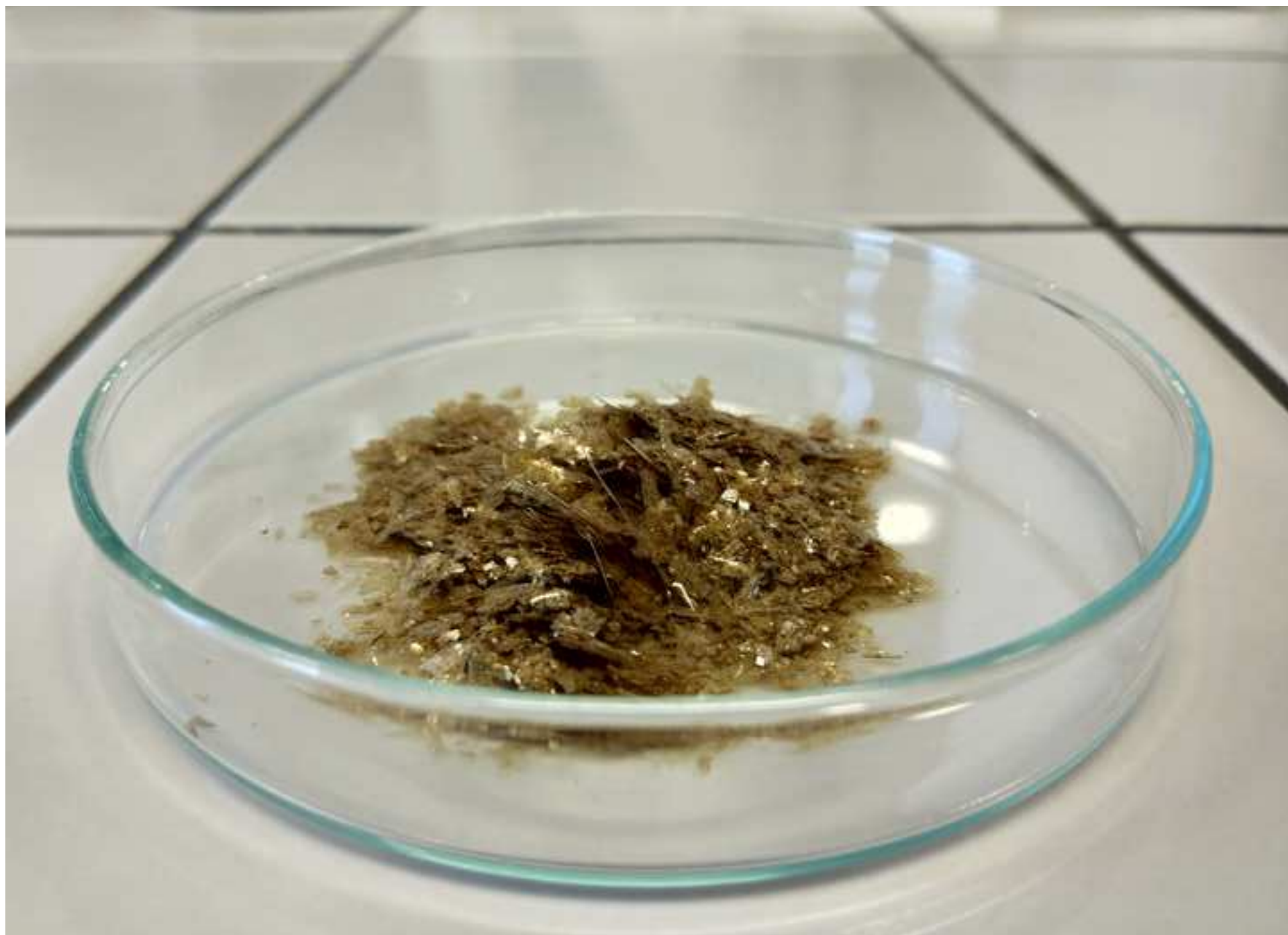
420 **REFERENCES:**

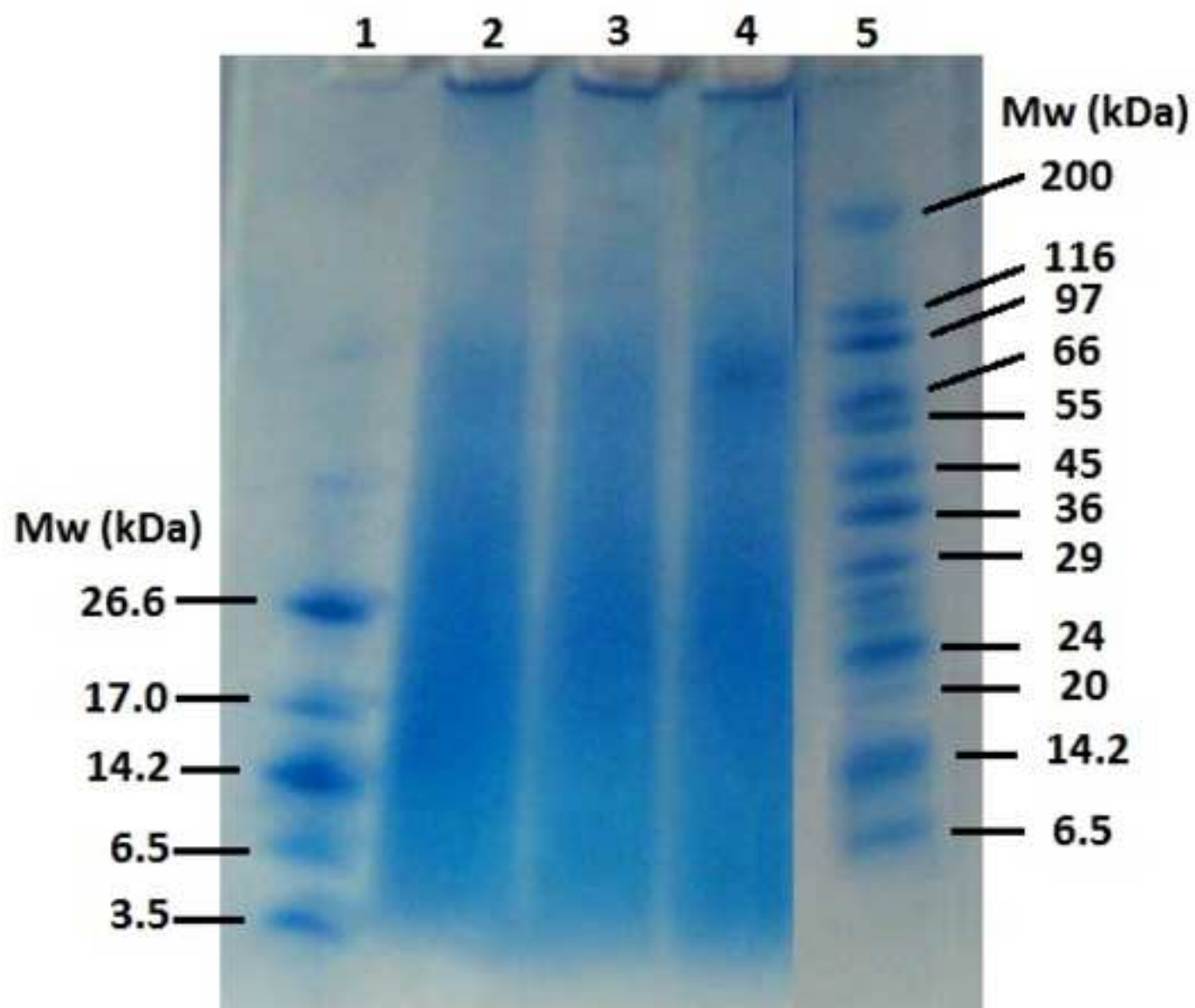
- 421 1. Poultry Slaughter, 2016 Summary. United States Department of Agriculture – National
422 Agricultural Statistics Services (February 2017).
- 423 2. McGovern, V. Recycling poultry feathers: more bang for the cluck. *Environ. Health Perspect.*
424 **108** (8), A336-A339, 10.1289/ehp.108-a366 (2000).
- 425 3. Gousterova, A., *et al.* Degradation of keratin and collagen containing wastes by newly
426 isolated thermoactinomycetes or by alkaline hydrolysis. *Lett. Appl. Microbiol.* **40** (5), 335-
427 340, 10.1111/j.1472-765X.2005.01692.x (2005).
- 428 4. Yamauchi, K., Yamauchi, A., Kusunoki, T., Khoda, A., Konishi, Y. Preparation of stable aqueous
429 solution of keratins, and physiochemical and biodegradational properties of films. *Biomed.*
430 *Mater. Res.* **31** (4), 439-444, 10.1002/(SICI)1097-4636(199608)31:4<439::AID-
431 JBM1>3.0.CO;2-M (1996).
- 432 5. Schrooyen, P.M.M., Dijkstra, P.J., Oberthur, R.C., Bantjes, A., Feijen, J. Partially
433 carboxymethylated feather keratins. 2. Thermal and mechanical properties of films. *J. Agric.*
434 *Food Chem.* **49** (1), 221-230, 10.1021/jf0004154 (2001).
- 435 6. Mark, H.F., Gaylord, N.G., Bikales, N.M. Encyclopedia of Polymer Science Technology: vol. 8:
436 Keratin to Modacrylic Fibers, New York, Wiley-Interscience (1968).
- 437 7. Bertsch, A., Cello, N. A biotechnological process for treatment and recycling poultry feathers
438 as a feed ingredient. *Bioresour. Technol.* **96** (15), 1703-1708, 10.1016/j.biortech.2004.12.026
439 (2005).

- 440 8. Grazziotin, A., Pimentel, F.A., de Jong, E.V., Brandelli, A. Nutritional improvement of feather
441 protein by treatment with microbial keratinase. *Animal Feed Sci. Technol.* **126** (1-2), 135-144,
442 <http://dx.doi.org/10.1016/j.anifeedsci.2005.06.002> (2006).
- 443 9. Brandelli, A. Bacterial keratinases: useful enzymes for bioprocessing agroindustrial wastes
444 and beyond. *Food Bioprocess Technol.* **1** (2), 105-116, 10.1007/s11947-007-0025-y (2008).
- 445 10. Gusta, R., Ramnani, P. Microbial keratinases and their prospective applications: an
446 overview. *Appl. Microbiol. Biotechnol.* **70** (1), 21-33, 10.1007/s00253-005-0239-8 (2006).
- 447 11. Vasileva-Tonkova, E., Gousterova, A., Neshev, G. Ecologically safe method for improved
448 feather wastes biodegradation. *International Biodeterior. Biodegradation* **63** (8), 1008-1012,
449 <http://dx.doi.org/10.1016/j.ibiod.2009.07.003> (2009).
- 450 12. Lodén, M. Hydrating Substance. In: Barel, A.O., Paye, M., Maibach, H.I. (eds). *Handbook*
451 *of Cosmetic Science and Technology*, New York: Informa, Healthcare, pp. 107–119 (2009).
- 452 13. Teglia, A., Secchi, G. Proteins in Cosmetics. In: Goddard, E.D., Gruber, J.V. (eds).
453 *Principles of Polymer Science and Technology in Cosmetics and Personal Care*. New York,
454 Marcel Dekker, Chapter 9 (1999).
- 455 14. Magdassi, S. Delivery systems in cosmetics. *Colloids and Surfaces A: Physicochem. Engin.*
456 *Aspects* **123-124**, 671-679, 10.1016/S0927-7757(97)03792-8 (1997).
- 457 15. Dahms, G., Jung, A. Method for producing a protein hydrolysate. U.S. Patent
458 20140323702 (2014).
- 459 16. Pons, R., Carrera, I., Erra, P., Kunieda, G, Solans, C. Novel preparation methods for highly
460 concentrated water-in-oil emulsions. *Colloids and Surfaces A: Physicochem. Engin. Aspects* **91**
461 (3), 259-266, [http://dx.doi.org/10.1016/0927-7757\(94\)02950-4](http://dx.doi.org/10.1016/0927-7757(94)02950-4) (1994).
- 462 17. Mokrejs, P., Hrnčirik, J., Janacova, D., Svoboda, P. Processing of keratin waste of meat
463 industry. *Asian J. Chem.* **24** (4), 1489-1494 (2012).
- 464 18. Mokrejs, P., Svoboda, P., Hrnčirik, J. Processing poultry feathers into keratin hydrolysate
465 through alkaline-enzymatic hydrolysis. *Waste Manage. Res.* **29** (3), 260-267,
466 10.1177/0734242X10370378 (2011).
- 467 19. Mokrejs, P., Krejci, O., Svoboda, P. Producing keratin hydrolysates from sheep
468 wool. *Orient. J. Chem.* **27** (4), 1303-1309 (2011).
- 469 20. Mokrejs, P., Krejci, O., Svoboda P., Vasek, V. Modeling technological conditions for
470 breakdown of waste sheep wool. *Rasayan J. Chem.* **4** (4), 728-735 (2011).
- 471 21. Polaskova, J., Pavlackova, J., Vltavska, P., Mokrejs, P., Janis, R. Moisturizing effect of
472 topical cosmetic products applied to dry skin. *J. Cosmet. Sci.* **64** (5), 329-340 (2013).
- 473 22. Polaskova, J., Pavlackova, J., Egner, P. Effect of vehicle on the performance of active
474 moisturizing substances. *Skin Res. Technol.* **21** (4), 403-412, 10.1111/srt.12206 (2015).
- 475 23. Verdier-Sévrain, S., Bonté, F. Skin hydration: a review on its molecular mechanisms. *J.*
476 *Cosmet. Dermatol.* **6** (2), 75-82, 10.1111/j.1473-2165.2007.00300.x (2007).
- 477 24. Darlenski, R., Sassning, S., Tsankov, N., Fluhr, J.W. Non-invasive in vivo methods for
478 investigation of the skin barrier physical properties. *Eur. J. Pharm. Biopharm.* **72** (2), 295-303,
479 10.1016/j.ejpb.2008.11.013 (2009).
- 480 25. Berardesca, E. EEMCO guidance for assessment of stratum corneum hydration: electrical
481 methods. *Skin Res. Technol.* **3** (2): 126–132, 10.1111/j.1600-0846.1997.tb00174.x (1997).
- 482 26. Rogiers, V. EEMCO guidance for the assessment of transepidermal water loss in
483 cosmetic sciences. *Skin Pharmacol. Appl. Skin Physiol.* **14** (2), 117–128, 56341 (2001).

- 484 27. Ali, S.M., Yosipovitch, G. Skin pH: from basic science to basic skin care. *Acta Derm.*
485 *Venerol.* **93** (3), 261-2177, 10.2340/00015555-1531 (2013).
- 486 28. Agache, P., Humbert, P. Measuring the Skin. Berlin, Springer-Verlag (2004).
- 487 29. International Ethical Guidelines for Biomedical Research Involving Human Subjects.
488 Council for International Organizations of Medical Sciences, Geneva (2002).
- 489 30. Ruland, J.K. Transdermal permeability and skin accumulation of amino acids. *Int. J.*
490 *Pharm.* **72** (2), 149–155, [http://dx.doi.org/10.1016/0378-5173\(91\)90053-Q](http://dx.doi.org/10.1016/0378-5173(91)90053-Q) (1991).
- 491 31. Draelos, Z.D. Therapeutic moisturizers. *Dermatol. Clin.* **18** (4), 597–607,
492 [http://dx.doi.org/10.1016/S0733-8635\(05\)70210-2](http://dx.doi.org/10.1016/S0733-8635(05)70210-2) (2000).
- 493 32. Information and Operating Instructions for the Multi probe Adapter MPA and its Probe.
494 Courage and Khazaka Electronic GmbH, Technical Charges (2013).







Name of Material/ Equipment	Company	Catalog Number
Material or chemicals		
LIPEX 100T	Novozymes	LJP30020
Savinase Ultra 16L	Novozymes	PXN40001
Potassium hydroxide, KOH	Sigma-Aldrich	302510289
Phosphoric acid solution, H ₃ PO ₄	Sigma-Aldrich	W290017
Sodium chloride physiological solution	Sigma-Aldrich	52455
Sodium hydroxide, NaOH	Penta s.r.o.	40216
AmiFarm (Cremor base-A)	Fagron	608425
Equipment		
IKA EUROSTAR POWER control-visc stirrers	IKA-labortechnik	Z404020
IKA Propeller stirrer, 3-bladed	IKA-labortechnik	R 1381
Dialysis tubing closures	Sigma-Aldrich	Z371017-10EA
Dialysis tubing cellulose membrane	Sigma-Aldrich	D9402-100FT
DOMO Pot with stailless, LCD	DOMO Elektronik	DO42325PC
Hettich zentrifugen Universal 32	Gemini bv	2770 GS1R
LT 3 shaking device	Fischer Scientific	6470.0002
KERN 440-47N	Kern	440-47N
KERN 770	Kern	770 -N
VENTICELL 222 - Komfort	BMT, MMM Group	C 131749
Vacucell 55 - EVO	BMT, MMM Group	B 050328
PULVERISETTE 19	Fritsch	19.1030.00
Multi Probe Adapter System MPA 5	Courage & Kazaka Electronic	10225237
Skin pH-meter PH 905 probe	Courage & Kazaka Electronic	
Corneometer CM 825 probe	Courage & Kazaka Electronic	
Tewameter TM 300	Courage & Kazaka Electronic	
Heidolph RZR 2020	Heidolph	13-225-007-03-1
Heidolph mechanical stirrer BR 10	Heidolph	Z336688-1EA
Fagor FS 12	Fagor	BTT-138
WTW bench pH/mV meter	WTW	Z313165

Container

RPC Superfos

Software

Microsoft Office 2010

Microsoft

C+K software

Courage and Khazaka Electronic GmbH

Comments/Description

Lipex® - enzyme produced by submerged fermentation of a genetically-modified microorganism, activity 100 KLU/g
Savinase® - enzyme produced by submerged fermentation of a genetically-modified microorganism, activity 16 KNPU-S/g
Potassium hydroxide, KOH, 97,0 %, Mr 56,11
Phosphoric acid solution, H₃PO₄, 85 wt. % concentration in water, Mr 98,00
Tablets of BioUltra NaCl physiological solution; 1 tablet in 1000 mL of water yields 0.9 % NaCl
Sodium hydroxide, NaOH, 97,0 %, Mr 40,00
Hydrophilic oil in water (O/W) cream base; the composition: aqua, paraffin, paraffin liquid, cetearyl alkohol, Laureth 4, sodium hydroxide, carbomer, methylparaben, propylparaben.

Digital laboratory stirrer, for tasks up to the high viscosity range, 230V, 1/cs
Propeller stirrer, 3-bladed, stirrer Ø: 45 mm, shaft Ø: 8 mm, shaft length: 350 mm
Dialysis tubing closures, red, size 110 mm
Dialysis tubing cellulose membrane, average flat width 76 mm (3.0 in.)
Preserving boiler stainless steel, 2000 W, 27-L container (diameter 37 cm, height 30 cm), temperature control 30-100 ° C, operation LC
Mid bench centrifuge, speed 18000 rpm
Orbital shaking device
Laboratory balance
Laboratory analytical balance
Drying oven, temperature control 30-100 ° C, air circulation control
Vacuum drying oven, temperature control 30-100 ° C
Universal cutting mill, rotor with V-cutting edges and fixed knives
MPA 5 Station - equipment for measurement hydration, TEWL and pH
Probe to specifically measure the pH on the skin surface or the scalp
Probe to determine the hydration level of the skin surface (Stratum corneum).
Probe for the assessment of the transepidermal water loss (TEWL)
Overhead stirrer, mechanical speed setting and stepless transmission; speed range 40-2000 rpm
Blade impeller crossed stirrer
Laboratory refrigerator with freezer space
High-performance bench pH and pH/conductivity meters for routine and high precision laboratory measurements in research or quality control laboratories

13-L plastic bucket, diameter 26 cm, height 26 cm

MPA 5 station operating software

D display

Table 1

Cosmetic formulation	Weight of ointment base [g]	Weight of keratin hydrolysate [g]
Ointment base	50	0
Ointment base + 2 % KH	49	1
Ointment base + 4 % KH	48	2
Ointment base + 6 % KH	47	3

Total weight [g]

50

Table 2

Men						
Time	1h	2h	3h	4h	24h	48h
Hydration (% change vs. Ointment base) ± SD						
Ointment base + 2 % KH	+16±15	+14±16	+12±9	+19±14	+11±18	+15±9
Ointment base + 4 % KH	+6±19	+1±18	+5±10	+7±16	+11±9	+14±15
Ointment base + 6 % KH	-3±25	-4±14	-7±18	-4±17	+11±14	-17±14
TEWL (% change vs. Ointment base) ± SD						
Ointment base + 2 % KH	-20±15	-20±22	-11±21	-20±21	-23±20	-21±17
Ointment base + 4 % KH	-28±12	-29±20	-28±20	-28±24	-47±20	-36±20
Ointment base + 6 % KH	-36±16	-41±21	-31±17	-36±17	-53±20	-54±17
pH						
Control	4.7±0.5	5.1±0.4	4.9±0.4	5.1±0.3	4.6±0.5	4.8±0.7
Ointment base	4.8±0.5	5.1±0.3	4.9±0.3	5.0±0.4	4.6±0.4	5.0±0.6
Ointment base + 2 % KH	5.0±0.6	4.8±0.4	4.9±0.5	4.9±0.5	4.7±0.3	5.0±0.6
Ointment base + 4 % KH	4.8±0.5	4.9±0.3	4.8±0.4	4.8±0.3	4.7±0.5	4.8±0.5
Ointment base + 6 % KH	4.7±0.5	5.0±0.2	4.9±0.4	4.8±0.4	4.8±0.6	5.0±0.6

Table 3

Women						
Time	1h	2h	3h	4h	24h	48h
Hydration (% change vs. Ointment base) ± SD						
Ointment base + 2 % KH	+22±7	+15±6	+15±8	+12±9	+14±14	+18±9
Ointment base + 4 % KH	-0.4±4	-6±5	-2±5	+1±7	+10±13	+15±10
Ointment base + 6 % KH	-12±5	-14±2	-9±7	-5±9	+8±12	+10±9
TEWL (% change vs. Ointment base) ± SD						
Ointment base + 2 % KH	-32±1.6	-16±3.0	-12±1.3	-20±0.9	-35±1.9	-38±1.6
Ointment base + 4 % KH	-41±1.1	-37±2.7	-24±0.8	-34±0.9	-44±1.5	-38±1.9
Ointment base + 6 % KH	-50±1.4	-39±2.2	-29±0.7	-39±0.9	-16±2.4	-33±2.1
pH						
Control	5.0±0.7	5.3±0.3	4.9±0.7	5.0±0.5	5.0±0.8	4.7±0.7
Ointment base	5.2±0.6	5.3±0.3	5.2±0.7	5.0±0.4	5.1±0.8	4.8±0.7
Ointment base + 2 % KH	5.4±0.7	5.1±0.4	4.9±0.4	5.1±0.7	4.9±0.7	5.0±1.0
Ointment base + 4 % KH	5.2±0.7	5.1±0.3	5.0±0.4	4.9±0.4	5.1±0.6	5.1±0.2
Ointment base + 6 % KH	5.2±0.7	5.2±0.2	5.0±0.4	5.0±0.3	5.4±0.6	5.2±0.4



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Author(s): P. MOKREJS, M. HUTTA, J. PAVLACKOVA, P. EGNER

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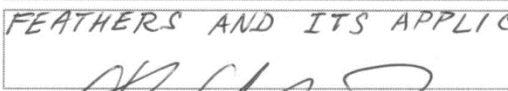
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Rebuttal Comments - JoVE56254

Dear Editor and Reviewers,

Below you will find comments referring to the revisions made to manuscript **JoVE56254 *Preparation of keratin hydrolysate from chicken feathers and its application in cosmetics.***

Revisions in the manuscript are highlighted in **red color**.

General Formatting:

1) JoVE is unable to publish manuscripts containing commercial sounding language.... Here are some examples from your manuscript - lipex, savinase, Excel 2010.

Revisions made in the manuscript:

„Lipex 100T“ was removed and general statement “lipolytic enzyme” is used

“Excel 2010” was replaced by “table processor”

“Savinase Ultra 16 L” was removed and general statement “proteolytic enzyme” is used

2) If you are re-using figures from a previous publication...

Answer: no figures from previous publications are included in the manuscript; all 3 figures are new ones.

3) Please copy-edit the entire manuscript for any grammatical errors you may find....

Answer: the language and grammar was reviewed.

4) Please provide email address of all authors on line 12

Answer: E-mail addresses of all co-authors were added.

5) Please add all the reagents and software to the Table of materials – cutting mill

Answer: software equipment (Microsoft Office 2010) and cutting mill (PULVERISETTE 19) are included in “JOVE_Materials” list.

Protocol:

1) Line 143 – was there any approval taken for use of animal products on humans? How were the individuals recruited? Please add the ethics statement here

Answer: There was no approval with respect to using animal products on humans. Keratin hydrolysate is a common cosmetic ingredient used in hair-care products (shampoos, conditioners etc.). Volunteers were recruited among employees and students of our university. The method of selecting was conducted according to “International Ethical Guidelines for Biomedical Research Involving Human Subjects. Council for International Organizations of Medical Sciences, Geneva (2002).” This is stated in last paragraph of „Introduction“.

2) Line 147 – what is the approximate weight of feathers?

Answer: the weight of feathers depends on the amount of processed feathers. In a laboratory experiment the weight of feathers was approximately 50 g

3) Line 160 – what sized container is used for this? 2-liter beaker, 10 liter bucket, etc?

Answer: a stainless steel 27-L container is used; this is added in the manuscript.

4) Line 161 – what is used for stirring?

Answer: an overhead stirrer is used; this is added in the manuscript

5) Line 166 – how do you measure the pH?

Answer: pH is measured with a laboratory WTW bench pH/mV meter; this equipment is included in the material/equipment list.

6) Line 168 – is the stirring done using an automated magnetic stirrer?

Answer: no, the stirring is done with overhead stirrer; this is added in the manuscript.

7) Line 171 – what is the sieve size?

Answer: the sieve size is 100 μm ; this is added in the manuscript.

8) Line 185 – what is the boiling point?

Answer: the boiling point was 100 $^{\circ}\text{C}$; this is added in the manuscript.

9) Line 185 – how do you heat the mixture?

Answer: the mixture was heated in the same stainless steel 27-L boiler container; this is added in the manuscript

10) Do you refer to the solution prepared in step 1.5 as KH? Please specify

Answer: yes. This is specified in the text.

11) Line 192 – what is the size of the container?

Answer: plastic bucket 26 cm in diameter and 26 cm high is used; this is added in the manuscript.

12) Line 194 – what is the volume of water used for dialysis?

Answer: the volumes are as follows: 400 mL of solution of KH against 4 L of distilled water.

13) Line 203 – label as note

Answer: it was labeled as note in the manuscript.

14) Line 204 – alcohol

Answer: the misspelling was corrected.

15) Line 207 – please provide detailed instructions of how to prepare these formulations. You may include a separate table

Answer: Commercial hydrophilic oil in water (O/W) cream base was used for preparing cosmetic formulations with keratin hydrolysate; this is added in the manuscript. Separate table (Table 1) referring to preparation of formulation is included in the manuscript.

16) Line 209 – please provide an example of how to achieve this 50 g final formulation

Answer: This is clearly specified in Table 1.

17) Line 213 – what do you mean by “condition them”?

Answer: Conditioning formulations for 2 h after withdrawing from ice-box means to warm them up to a room temperature to simulate common usage of cosmetic formulations.

18) Line 218 – lable as note

Answer: it was labeled as note in the manuscript.

19) Line 231 and rest of the protocol – SI units mL

Answer: in all protocol “mL” instead of “ml” is used.

20) Notes should start on a new line. Do not highlight notes.

Answer: all notes start on a new line and are not highlighted.

Results:

1) line 249 – please provide a representative picture of KH

Answer: Figure 2 showing representative picture of keratin hydrolysate is added in the manuscript.

2) line 259 – how many men and women volunteers?

Answer: 10 men and on 10 women; this is added in the manuscript.

Figure legends:

1) Expand the figure legends, add information about number of volunteers, presentation of values [SD or SEM], was any statistical analysis performed?

Answer: required information is added. Readings we processed by basic numerical characteristics of the descriptive statistics. SD is calculated.

Discussion:

1) Please expand the discussion so that it covers the following in detail and in paragraph form: 1) modifications and troubleshooting, 2) limitations of the technique, and 3) critical steps within the protocol.

Answer: The discussion was expanded as suggested in paragraph forms.

2) Is it possible to provide long term analysis – 7 days or longer? If not, insert comments here

Answer: In “Discussion” there is a statement “Longer-term analysis (more than 2 days) was not accomplished.”

Figures:

1) The difference between control, OB and other groups is not clear based on figure 1. will it be possible to include high resolution images for each group?

Answer: New Figure 1 with high resolution images for each group is included.

Reviewers' comments:

Reviewer #1:

Minor Concerns:

- The inclusion of electrophoresis or chromatography data showing the MW distribution would be very useful.

Answer: Gel electrophoresis data (SDS-PAGE) showing the molecular weight distribution of KH is shown in Figure 3.

- lines 160 - 162: The activity of the enzyme should be considered when determining how much to use. (related to weight of dry feathers)

Answer: the activity of enzyme is 100 KLU/g; this is added in the manuscript. The activity is added to “JOVE_Materials” list as well. Note: KLU = kilo lipase unit

- line 197: Give some suggestion on ratio of solution volume to plate area.

Answer: ratio of solution volume to plate area is 500 mL / 1,000 cm²

- lines 212 - 213: Are there other options if a digital stirrer is not available? With nondigital stirrer, how can one estimate the correct speed, and how important is it?

Answer: Homogenizing the mixture of ointment base with keratin hydrolysate can be done with non-digital stirrer as well. On non-digital stirrer there are scales with approximate speed (in rpm) as well. Gentle stirring is the best way how to do this step.

- line 339: Where is the evidence for the broad MW distribution?

Answer: This is obvious from Figure 3 „SDS-PAGE of keratin hydrolysate and protein standards“ which is included in manuscript.

- line 340: What is the basis for "We believe".

Answer: The basis is from the broad MW distribution of keratin hydrolysate. The higher values for

hydration of the skin can be assigned to low molecular weight fractions of keratin hydrolysate penetrating to the epidermis and helping to bind water in epidermis layer. On the other hand, higher molecular weight fractions of keratin hydrolysate create thin film on the outer layer of epidermis and thus enhance barrier properties of epidermis which leads to lower trans-epidermal water loss. It might be better to replace “We believe” by “We suggest” – this is corrected in the manuscript.

Additional Comments to Authors:

The manuscript can be understood, but is not easily readable for several reasons. Please edit the language. Some suggestions, not all inclusive:

- lines 49-50: The numbers are too precise for what are really estimates, try 62.5 billion, etc.

Answer: The numbers were corrected.

- lines 59 and 61: mixed US and UK English usage in disulfide and sulphitolysis

Answer: The text was corrected for US English only.

- line 148: Place the feathers on a flat plate and dry overnight at 50 °C.

Answer: Our previous sentence was replaced by this suggested one.

- lines 192-193: Dialyze keratin hydrolysate using 12K MWCO membrane to remove small peptides and salts.

Answer: Our previous sentence was replaced by this suggested one.

- lines 224 - 225: Apply two strips to the inner side of the right forearm, and three to the inner side of the left forearm.

Answer: Our previous sentence was replaced by this suggested one.

- line 236: readings at 1, 2, 3, 4, 24 and 48 hours would be 6 points, make it clear that 5 readings are taken at each site and each interval.

Answer: The sentence was revised to make it clear as suggested.

- In several place, explanations are too wordy. Proof read the manuscript carefully, watch for American grammar, agreement of nouns and verbs with respect to singular and plural, consistency of verb tense, and spelling, especially of author's names in the reference list.

Answer: The manuscript was proofed again.

Reviewer #2:

Major Concerns:

- The pollution of hydrolysis with alkaline pretreatment would be a imperfection.

Answer: After hydrolysis keratin hydrolysate is dialyzed to remove salts (and small peptides) so there will be no problem with alkaline pollution. In addition, pH of 1.0% solution of KH is 5.3 and fulfils requirements for cosmetic-grade hydrolysates. This is added in the manuscript.

Minor Concerns:

- If there is fluorescence of the keratin and it may have some effect for sun screen.

Answer: The effect of keratin hydrolysate for sun screen is the matter of further research.

Additional Comments to Authors:

- The mould proof may need for the keratin solution.

Answer: The mould proof for the keratin solution is not included. After hydrolysis, KH solution was heated to the boiling point to inactivate enzyme and to sterilize it. Then, KH solution is dried and in further steps dry powder is used.